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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/687,541	10/16/2003	David A. Morgenstern	MTC 6888.2 (39-21 (52925)	7748
321 SENNIGER PO	7590 06/14/2007 OWERS	EXAMINER		
	ONE METROPOLITAN SQUARE ECHELMEYER, ALIX ELIZABETH			
16TH FLOOR ST LOUIS, MO			ART UNIT	PAPER NUMBER
			1745	
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			NOTIFICATION DATE	DELIVERY MODE
			06/14/2007	ELECTRONIC

Please find below and/or attached an Office communication concerning this application or proceeding.

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	Application No.	Applicant(s)				
	10/687,541	MORGENSTERN, DAVID A.				
Office Action Summary	Examiner	Art Unit				
	Alix Elizabeth Echelmeyer	1745				
The MAILING DATE of this communication app Period for Reply	pears on the cover sheet with	the correspondence address				
A SHORTENED STATUTORY PERIOD FOR REPL WHICHEVER IS LONGER, FROM THE MAILING D - Extensions of time may be available under the provisions of 37 CFR 1.4 after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period - Failure to reply within the set or extended period for reply will, by statute Any reply received by the Office later than three months after the mailine earned patent term adjustment. See 37 CFR 1.704(b).	NATE OF THIS COMMUNICA 136(a). In no event, however, may a repl will apply and will expire SIX (6) MONTH e, cause the application to become ABAN	ATION. ly be timely filed IS from the mailing date of this communication. NDONED (35 U.S.C. § 133).				
Status		•				
1)⊠ Responsive to communication(s) filed on <u>04 A</u>	April 2007.					
2a) This action is FINAL . 2b) ∑ This	This action is FINAL . 2b)⊠ This action is non-final.					
•—	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is					
closed in accordance with the practice under	Ex parte Quayle, 1935 C.D. 1	11, 453 O.G. 213.				
Disposition of Claims						
4) Claim(s) 1-52,88-108 and 117-130 is/are pend	4) Claim(s) <u>1-52,88-108 and 117-130</u> is/are pending in the application.					
4a) Of the above claim(s) <u>53-87 and 109-116</u> i	is/are withdrawn from conside	eration.				
5) Claim(s) is/are allowed.	•					
6)⊠ Claim(s) <u>1-52,88-108 and 117-190</u> is/are reject	cted.					
7) Claim(s) is/are objected to.						
8) Claim(s) are subject to restriction and/o	or election requirement.					
Application Papers						
9) The specification is objected to by the Examine	er.					
10)☐ The drawing(s) filed on is/are: a)☐ acc	cepted or b)□ objected to by	the Examiner.				
Applicant may not request that any objection to the	•					
Replacement drawing sheet(s) including the correct 11) The oath or declaration is objected to by the E						
Priority under 35 U.S.Ç. § 119						
12) Acknowledgment is made of a claim for foreign	n priority under 35 U.S.C. & 1	119(a)-(d) or (f)				
a) ☐ All b) ☐ Some * c) ☐ None of:	in priority under 55 0.5.6. § 1	19(a)-(d) 01 (1).				
1. Certified copies of the priority documents have been received.						
3. Copies of the certified copies of the price						
application from the International Burea	au (PCT Rule 17.2(a)).					
* See the attached detailed Office action for a list	t of the certified copies not re	eceived.				
Attachment(s)	_					
 Notice of References Cited (PTO-892) Notice of Draftsperson's Patent Drawing Review (PTO-948) 	4) Interview Sur Paper No(s)/	mmary (PTO-413) Mail Date				
3) Information Disclosure Statement(s) (PTO/SB/08)	5) Notice of Info	ormal Patent Application				
Paper No(s)/Mail Date	6)	÷				

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DETAILED ACTION

Election/Restrictions

- 1. Claims 53-87 and 109-116 are withdrawn from further consideration pursuant to 37 CFR 1.142(b) as being drawn to a nonelected group, there being no allowable generic or linking claim. Election was made **without** traverse in the reply filed on April 4, 2007.
- 2. In the Election of Species requirement, Applicants elected nickel as the base metal material.
- 3. Claims 1-52, 88-108 and 117-130 are pending and are rejected for the reasons given below.

Claim Interpretation

The product-by-process limitations of claims 1, 4-6, 9-11, 33-35, 39-41, 88-90, 101, 106-108, 117-119 and 127 are not given patentable weight since the courts have held that patentability is based on a product itself, even if the prior art product is made by a different process (see <u>In re Thorpe</u>, 227 USPQ 964, (CAFC 1985), <u>In re Brown</u>, 173 USPQ 685 (CCPA 1972), and <u>In re Marosi</u>, 218 USPQ 289, 292-293 (CAFC 1983)).

Claims 1, 88-90, 101, 106-108, 117-119 and 127 are drawn to methods for forming the catalyst. While the claims are process claims, the process of those claims is

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that of reforming a feed gas to produce hydrogen, and not the process of forming the catalyst.

Additionally, claims 4-6, 9-11, 33-35 and 39-41 are drawn to the method by which the surface area is measured, specifically the Brunauer-Emmett-Teller method. That method is not given patentable weight since the surface area limitations are met.

Claim Rejections - 35 USC § 103

- 5. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 6. Claims 1-52, 88-108 and 117-130 are rejected under 35 U.S.C. 103(a) as being unpatentable over Marino et al. (Hydrogen from Steam Reforming of Ethanol) in view of Sargent (US Patent 2,892,801).

Marino et al. teach a catalyst made of copper on the surface of a porous alumina substrate (abstract; p. 1095 2nd column).

Regarding claims 2, 20, 94 121 and 122, the catalyst is used to reform ethanol to produce hydrogen (Introduction).

As for claims 3, 21, 32 and 50, Marino et al. teach that the hydrogen produced may be used in a fuel cell. It would have been obvious to one having ordinary skill in the

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art to provide hydrogen and oxygen to a fuel cell, since both components are needed for the fuel cell to produce electricity.

With regard to claims 4-6, 9-11, 33-35 and 39-41, Marino et al. teach that the surface area of copper in the catalyst may be 98.80 m²/g (Table 4. p. 1099). It is the position of the examiner that 98.80 is about 80. As for the method by which the surface area is measured, as discussed above, the method by which the surface area is measured is not given patentable weight since the surface area limitations are met.

As for claims 22-24, 28 and 123, Marino et al. teach that the reaction is carried out at 300°C (p. 1096 column 1).

Regarding claims 1, 12, 25-27, 38, 42, 51, 52, 88-93, 101-108 and 117-120, and 127-130, Marino et al. fail to teach that the substrate is a metal, specifically nickel.

Marino et al. do teach that the presence of nickel is desirable in the catalyst, since the presence of nickel tends to ensure that copper will stay on the surface of the substrate (p. 1097 2nd column) and that the addition helps to maximize hydrogen production (p. 1-99 1st column).

Sargent teaches a catalyst made of a copper-plated nickel sponge (column 1 lines 40-44). Sargent further teaches that the catalyst may be used in dehydrogenation (column 4 lines 44-47).

As for claims 7, 8, 15, 16, 36, 37, 45 and 46, Sargent teaches that the final catalyst may be from 0.5 percent to 75 percent by weight copper (column 2 lines 21-24). Specific examples 5 and 6 on columns 3 and 4 teach that the catalyst may be as much as 27.0 or 44.4 percent by weight copper.

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Regarding claims 13, 14, 43 and 44, Example 1 of Sargent teaches a catalyst of 9.5 percent by weight copper (column 3). This would yield a catalyst of 90.5 percent by weight nickel.

As for claims 95, 98 and 124, since the catalyst, reactant, and temperature are the same as in the presently claimed invention, the reaction would inherently produce methane.

Regarding claims 96, 97, 99, 100, 125 and 126 it is well known to run internal combustion engines on gaseous fuels such as methane and hydrogen (see, for example, US Patent 5,398,663 column 1 lines 20-24).

With regard to claims 17-19 and 47-49, since the catalyst structure, including the amount and surface area of the components, is identical to the presently claimed invention, the amount of nickel at the surface of the catalyst would inherently be the same in the combination of Marino et al. and Sargent as in the instantly claimed invention.

With regard to claims 29-31, since the catalyst structure, including the amount and surface area of the components, is identical to the presently claimed invention, the thermal conductivity of the catalyst would inherently be the same in the combination of Marino et al. and Sargent as in the instantly claimed invention.

It would be desirable to use the copper plated nickel sponge of Sargent as the catalyst in the reaction of Marino et al. since Marino et al. teach that the presence of nickel improves the performance of a copper catalyst.

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Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use the copper plated nickel sponge of Sargent as the catalyst in the reaction of Marino et al. since Marino et al. teach that the presence of nickel improves the performance of a copper catalyst.

Response to Arguments

7. Applicant's arguments, see Remarks, filed December 7, 2006, with respect to the rejections of claims have been fully considered and are persuasive. Therefore, the rejection has been withdrawn. However, upon further consideration, a new ground of rejection is made, see above.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Alix Elizabeth Echelmeyer whose telephone number is 571-272-1101. The examiner can normally be reached on Mon-Fri 7-4:30.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's trainer, Susy N. Tsang-Foster can be reached on 571-272-1293. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Alix Elizabeth Echelmeyer Examiner Art Unit 1745

aee

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